गुरू घासीदास विश्वविद्यालय (केंद्रीय विश्वविद्यालय) कोनी, बिलासपुर - 495009 (छ.ग.)



Guru Ghasidas Vishwavidyalaya (A Central University Established by the Central Universities Act 2009 No. 25 of 2009) Koni, Bilaspur – 495009 (C.G.)

# **List of Revised Courses**

**Department** : *Chemical Engineering* 

Program Name : **B.Tech.** 

Academic Year : 2019-20

# **List of Revised Courses**

Sr. No.	Course Code	Name of the Course
01.	CH03TBS06	Mathematics-III
02.	CH03TPC01	Material And Energy Balance Calculations
03.	CH03TPC02	Fluid Mechanics
04.	CH03TPC03	Thermodynamics-II
05.	CH04TPC04	Numerical Methods In Chemical Engineering
06.	CH04TPC06	Particle And Fluid Particle-Processing
07.	CH04TPC07	Process Instrumentation

गुरू घासीदास विश्वविद्यालय (केट्रीय विस्तविद्याल अधिनियम 2009 क. 25 के अंतर्गत स्वापित केन्द्रीय विश्वविद्यालय) कोनी, बिलासपुर - 495009 (छ.ग.)



Guru Ghasidas Vishwavidyalaya (A Central University Established by the Central Universities Act 2009 No. 25 of 2009) Koni, Bilaspur – 495009 (C.G.)

## Minutes of Meetings (MoM) of Board of Studies (BoS)

# Academic Year : 2019-20

School : School of Studies of Engineering and Technology

**Department** : *Chemical Engineering* 

Date and Time : *May 13, 2019 – 11:30 AM* 

Venue : HoD Room

#### Minutes of Meetings

The scheduled meeting of member of Board of Studies (BOS) was held today (13<sup>th</sup> May 2019) in the office of Prof. S.N. Saha, Chemical Engineering Department to discuss the B.Tech. III and IV semester scheme and syllabi. The following members were present in the meeting:

- 1. Prof. S. N. Saha (Member BOS, Dept. of Chemical Engg)
- 2. Dr. Anil Kumar Chandrakar (HoD (l/c), Chemical Engg. -cum- Chairman, BOS)
- Mrs. A. N. Joshi (Member BOS, Dept. of Chemical Engg)
- 4. Mr. Amit Jain (Invited Member, Astt. Prof., Dept. of Chemical Engg.)
- 5. Mr. G. P. Dewangan (Invited Member, Astt. Prof., Dept. of Chemical Engg.)
- 6. Dr. Sandeep Singh (Invited Member, Astt. Prof., Dept. of Mathematics)
- Dr. Sandeep Singh (Invited Member, Astr. Prof., Dept. of Mechanical Engg.)
   Ms. Jasinta Poonam Ekka (Invited Member, Astr. Prof., Dept. of Mechanical Engg.)

The committee discussed and resolved to approve the scheme and syllabi of B.Tech. III and IV semester.

Since external expert BOS member Prof. (Mrs.) A. B. Soni, Dept. of Chemical Engg. and external industry expert BOS member, Mr. Suprangya Mohanty could not attend this meeting due to their pre-occupation, as per discussion between members present in the meeting, this scheme and syllabus is being sent to external BOS members for their review and formal

consent Prof. S. N. Saha

Member, BOS Professor, Chemical Engg.

Mr.

Invited Member

Dr. Anil Kumar Chandrakar Chairman, BOS HoD (I/e), Chemical Engg.

Mrs. A. N. Joshi Member, BOS Astt. Prof., Chemical Engg.

Mr. G. P. Dewangan Invited Member Astt. Prof., Chemical Engg.

Astt. Prof., Chemical Engg.

Dr. Sandeep Singh Invited Member Astt. Prof.,Dept. of Mathematics

Ms. Jakinta Poonam Ekka Invited Member Astt. Prof., Mechanical Engg.

गुरू घासीदास विश्वविद्यालय (केंद्रीय विस्तविद्याल अधिनियम 2009 क. 25 के अंतर्गत स्वापित केंद्रीय विश्वविद्यालय) कोनी, बिलासपुर - 495009 (छ.ग.)



Guru Ghasidas Vishwavidyalaya (A Central University Established by the Central Universities Act 2009 No. 25 of 2009) Koni, Bilaspur – 495009 (C.G.)

The following courses were revised in the of B. Tech. Second year (III and IV Semesters) :

- Mathematics-III (CH03TBS06)
- Material And Energy Balance Calculations (CH03TPC01)
- Fluid Mechanics (CH03TPC02)
- Thermodynamics-II (CH03TPC03)
- Numerical Methods In Chemical Engineering (CH04TPC04)
- Particle And Fluid Particle-Processing (CH04TPC06)
- Process Instrumentation (CH04TPC07)

The following new courses were introduced in the of B. Tech. Second year (III and IV Semesters):

- Biology (CH03TBS05)
- Engineering And Solid Mechanics (CH03TES04)
- Process Instrumentation Lab (CH04PPC05)

Mandraka

विभागार्ध्यक्ष, रासायनिक अभियांत्रिकी HoD, Chemical Engineering प्राद्योगिकी संस्थान/Institute of Technology गुरू घासीदास विश्वविद्यालय, बिलासपुर (छ.ग.) Guru Ghasidas Vishwavidyalaya, Bilasour (C.G.)

Signature & Seal of HoD

### गुरू घासीदास विश्वविद्यालय (केंद्रीय विश्वविद्यालय) कोनी, बिलासपुर - 495009 (छ.ग.)



Guru Ghasidas Vishwavidyalaya (A Central University Established by the Central Universities Act 2009 No. 25 of 2009) Koni, Bilaspur – 495009 (C.G.)

### **Scheme and Syllabus**

#### SCHOOL OF STUDIES, ENGINEERING & TECHNOLOGY GURU GHASIDAS VISHWAVIDYALAYA, BILASPUR (C.G.) (A Central University Established by the Central University Ordinance 2009, No. 3 of 2009) SCHEME FOR EXAMINATION (Effective from session 2019-20) B. TECH. (FOUR YEAR) DEGREE COURSE, CHEMICAL ENGINEERING SECOND YEAR, THIRD SEMESTER Course No. Periods **Evaluation Scheme** S.No. Subject Sessional Exam Credits т Р L THEORY IA ESE Total 01 CH03TBS05 Biology New Course 3 1 0 30 70 100 4 02 CH03TBS06 Mathematics -III 3 1 0 30 70 100 4 03 CH03TES04 Engineering and Solid Mechanics New Course 3 1 0 30 70 100 4 04 CH03TPC01 Material and Energy Balance Calculations 3 T. 0 30 70 100 4 05 CH03TPC02 Fluid Mechanics 3 1 30 0 70 100 4 06 CH03TPC03 Thermodynamics-II 3 1 0 30 70 100 4 Practical 01 CH03PPC01 Chemical Engineering Lab-I 0 0 3 30 20 50 1.5 02 CH03PPC02 Fluid Mechanics Lab 0 0 3 30 20 50 1.5 Total 18 6 6 700 27 IA -- Internal Assessment **ESE- End Semester Examination** Total Marks -700 Total Periods - 30 Total Credits - 27 BOS held on 13th May 2019 SCHOOL OF STUDIES, ENGINEERING & TECHNOLOGY GURU GHASIDAS VISHWAVIDYALAYA, BILASPUR (C.G.) (A Central University Established by the Central University Ordinance 2009, No. 3 of 2009) SCHEME FOR EXAMINATION (Effective from session 2019-20) B. TECH. (FOUR YEAR) DEGREE COURSE, CHEMICAL ENGINEERING SECOND YEAR, FOURTH SEMESTER Course No. Periods **Evaluation Scheme** S.No. Subject Sessional Exam Credits L Т P THEORY IA ESE Total Business Communication and Presentation 01 CH04THS02 3 0 0 30 70 100 3 Skill 02 CH04TPC04 Numerical Methods in Chemical Engineering 3 1 0 30 70 100 4 03 CH04TPC05 Inorganic Chemical Technology 3 0 0 30 70 100 3 04 Particle and Fluid Particle-Processing CH04TPC06 3 0 0 30 70 100 3 05 CH04TPC07 Process Instrumentation 3 0 0 30 70 100 3 Practical Numerical Methods in Chemical Engineering 01 CH04PPC03 0 0 3 30 20 50 1.5 Lab 02 CH04PPC04 Particle and Fluid Particle-Processing Lab 0 0 3 30 20 50 1.5 03 CH04PPC05 New Cours Process Instrumentation Lab 0 0 3 30 20 50 1.5 Total 15 1 9 650 20.5 IA - Internal Assessment **ESE- End Semester Examination** Total Marks -650 Total Periods - 25 Total Credits - 20.5 BOS held on 13<sup>th</sup> May 2019 Gall

**Program Revision** 

Criteria – I (1.1.2)



**B.** Tech Syllabus

### Department of Chemical Engineering

#### CH03TBS06 Mathematics-III

[L:3, T:1,P:0] 70% Change

Unit I : Introduction to statistics, mathematical statistics, variable, frequency distribution, exclusive and inclusive class intervals, type of series, graphical representation: histogram, frequency polygon, ogive measure of central tendency various types of averages, Mean median mode for grouped and ungrouped data, geometric mean, harmonic mean, measure of dispersion Skewness and Kurtosis.

Unit II : Curve fittings by method of least square- straight line parabola correlation-scatter cliagrem's Karl Pearson's coefficient of correlation. Limits for correlation coefficient, rank correction. Regression linear regression, equation to the line of regression. Regression coefficient, angle between two lines of regression.

Unit III: Theory of probability-Mathematical and statistical definition of probability sample space. Finite sample space sample point, events theorem of total probability. Sample and compound event. Conditional probability, theorem of compound probability, Baye's theorem, use of binomial theorem.

Unit IV : Theroritical distribution- Binominal distribution mean, standard deviation and Pearson's  $\beta$  and  $\gamma$  coefficient. Poisson distribution, mean, variance normal distribution.

Unit V : Random and simple sampling-mean, and standard deviation in simple sampling of attribute, test of significant for large sample test of significance based on Chi square, T, F and Z distribution degree of freedom, condition for applying,

#### Text Books:

- 1. M. Roy, "Mathematical Statistics"
- 2. Biswal, "Probability & Statistics", PHI.

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- 3. A.A.AFTI, "Statistics analysis"
- 4. S. C. Gupta and Kapoor, "Fundamental of Mathematical Statistics"



B. Te	ch Syllabus	Dep	partment of Chemical Engineer
CHO	3TPC01	Material and Energy Balanc	e Calculations [L:3, T:1,P:
Ohie	ctives		
		as a basis for all further chemical	anginogring serveres that
the cu	urriculum.	a a close for an ranner chemical	
			50% Change
Conte			
1.	Introductory e	oncepts of units, physical qua	antities in chemical engineerin
	officiationiess g	roups, "pasis" of calculations	[21 + 171]
4.	Gases, Vapours	and Liquids: Equations of state, V	apour pressure, Clausius-Clapeyr
3	Humidity and S	hart, Duhring's plot, Raoult's law,	[6L+2T]
	its use.	aturation, humid heat, humid volu	
4.		es with recycle, bypass and purge.	[6L+2T]
		1. S.	·
5.	Material Balan	ce: With chemical reaction, Co	ncept of stoichiometry and me
	balances, examp	les, including combustion.	[6L+2T]
2	Material Dat		
0.	reaction.	e: Introduction, solving material	
	reaction,		[6L+2T]
7.	Energy balance	open and closed system, heat	conscitut colculation of anthal
	changes.	open and there openin, near	[6L+2T]
8,	Energy balance	s with chemical reaction: Heat	of reaction, Heat of combustic [6L+2T]
Total	[45L+15T]		
Sugge	sted Text Books		
1.	S. N. Saha, "Ch	emical Process Engineering Calc	ulation? Diverset Dei Dell'
	Co. (Pvt.) Ltd., N	New Delhi	diation , Dhanpat Kai Publicatio
2.	Bhatt, B. I., V	ora, S. M., "Stoichiometry", For	urth Edition. Tata McGraw H
	Publishing Com	bany Ltd, 2004.	Landon vola moendw fi
Sugge	sted References I	Books	
1.	Felder, R. M.; R.	ousseau, R. W., "Elementary Princi	ples of Chemical Processes", Thi
	Edition, John Wi	ley & Sons, 2000	
4.	Material & Enor	Watson, K. M., Ragatz, R. A., "C	hemical Process Principles, Part
3.	Himmelblau, D	y Balances", Second Edition, CBS M., Riggs, J. B. "Basic Princip	b rublishers & Distributors, 2004
-37.89	Engineering", E	ighth Ed., Pearson India Education	Services 2015
4.	Venkataramani,	V., Anantharaman, N., Begum,	K. M. Meera Sheriffa."Proce
	Calculations", S	econd Edit ion, Prentice Hall of In	dia.
5.	Sikdar, D. C., "C	hemical Process Calculations", Pr	entice Hall of India.
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B. Tech Syllabus

#### CH03TPC02 Fluid Mechanics

# Department of Chemical Engineering [L:3, T:1,P:0]

33% Change

#### Objectives:

The objective of this course is to introduce the mechanics of fluids (fluid statics and fluid dynamics), relevant to Chemical Engineering operations. The course will introduce students to forces on fluids, hydrostatic forces on submerged bodies, Eulerian and Lagrangian descriptions of flow, flow visualization, integral analysis involving mass and momentum balances, Bernoulli equation, flow through pipes and ducts, flow measurement and instruments, flow transportation - pumps, blowers and compressors, conservation of mass, linear and angular momentum in differential form, Navier-Stokes equation, viscous flows, skin and form friction, potential flows and boundary layer theory. Turbulence and turbulent flows will be introduced.

Contents :

onic		
1.	Introduction to fluids, Types of fluids, Concept of viscosity, Forces on	fluids, Normal
	and shear stresses.	[3L+1T]
2.	Fluid statics -Hydrostatic equilibrium, pressure distribution, Manome	try, Forces on
	submerged bodies, Buoyancy,	[3L+1T]
3.	Kinematics of fluid flow- Eulerian and Lagrangian descriptions, Flow	visualization.
	Streamfunction, Vorticity and Circulation.	[3L+1T]
4.	System and control volume approaches, Integral balances - mass ar	
	Euler's equation of motion, Bernoulli equation and applications.	[4L+2T]
5.	Reynolds number, Laminar Flow for Newtonian and Non Newtonian fl	
	flow through pipes and close channels and its characteristic equations.	[5L+1T]
6.	Head loss in pipe flow, Friction losses due to sudden changes in veloci	
	of flow, expansion, contraction, Effect of fittings.	[6L+2T]
7.	Flow measurement, variable head meters, variable area meter, insertion	
	tion invision entering variable neural inviens, variable area ineter, insertion	[3L+1T]
8	Transportation of fluids - pumps, blowers, compressors selection :	
	pumps,	
0	Differential analysis: mass and momentum balances, Navier-Sto	[3L+1T]
24		
	Unidirectional flow, Viscous flow, Stokes law, Skin drag and pressure of	T
10	Deputing the state of the state	[6L+2T]
	Potential flow, Potential function, Solution of Laplace equation.	[3L+1T]
11	Boundary layer theory, Blasius solution, Boundary layer separation.	[6L+2T]
		+
	[45L + 15T]	
gge	sted Text Books	

- 1. M. Whit e, Fluid Mechanics, 8 th Edition, Tata-McGraw Hill, 2016.
- V. Gupta and S. K. Gupta, Fundamentals of Fluid Mechanics, 2<sup>nd</sup> Edition, New Age International 2011.
- W. L. McCabe, J. C. Smith and P. Harriot, Unit Operations of Chemical Engineering, 7<sup>th</sup> Edition, McGraw-H il 1 International Edition 2005.

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4. O. Wilkes, Fluid Mechanics for Chemical Engineers, Prentice Hall of India, 2005

BOS held on 13th May 2019 12. All Sugar

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<ul> <li>CH03TPC03 Thermodynamics-II</li> <li>Pre-requisites: Thermodynamics-I</li> <li>Objectives:</li> <li>To introduce the concepts of fugacity, activity coefficeaction equilibrium. Introduction to molecular thermodynamics:</li> <li>Contents</li> <li>Review of first and second law of thermodynamic</li> <li>Solution Thermodynamics: fundamental properties, definition of pure species and species in solution, the ideal sol</li> <li>Liquid phase properties from VLE, Models for oproperty change on mixing.</li> <li>Vapor-liquid equilibrium: phase rule, simple Raoult's law; VLE from K-value correlations; FI</li> <li>Ideal solutions, activity and activity coefficient UNIQUAC models.</li> <li>Liquid-Liquid Equilibria; Vapor-Liquid-Liquid Solid-Gas equilibria.</li> <li>Chemical reaction equilibria: equilibrium criter of equilibrium constant at different temperatur react ions, multi reaction equilibria.</li> <li>Total [45L+15T]</li> <li>Suggested Text Books</li> <li>J.M. Smith, H.C. Van Ness and M.M. Abbott, "In Thermodynamics", 7th edition, McGraw-Hill Int 2, Y.V.C.Rao, "Chemical Engineering Thermodynamics", 7th edition, McGraw-Hill Int 2, Y.V.C.Rao, "Chemical Engineering Thermodynamics" Prentice Hall Of India, New Delhi 2011</li> </ul>	[L:3, T:1,P:0] % Change	
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<ul> <li>Solid-Gas equilibria.</li> <li>7. Chemical reaction equilibria: equilibrium criter of equilibrium constant at different temperatur react ions, multi reaction equilibria.</li> <li>Fotal [45L+15T]</li> <li>Suggested Text Books <ol> <li>J.M. Smith, H.C. Van Ness and M.M. Abbott, "In Thermodynamics", 7th edition, McGraw-Hill Int</li> <li>Y.V.C.Rao, "Chemical Engineering Thermodyna Hyderabad, 1997.</li> <li>K V Narayanan, "A Textbook of Chemica Prentice Hall Of India, New Delhi 2011</li> </ol> </li> <li>Suggested References Books <ol> <li>R.C. Srivastava, "Thermodynamics an core comparison of the section of th</li></ol></li></ul>	nt, Wilson, NRTL, UNIFA [6L+	
<ul> <li>of equilibrium constant at different temperature react ions, multi reaction equilibria.</li> <li>Fotal [45L+15T]</li> <li>Suggested Text Books <ol> <li>J.M. Smith, H.C. Van Ness and M.M. Abbott, "In Thermodynamics", 7th edition, McGraw-Hill Int</li> <li>Y.V.C.Rao, "Chemical Engineering Thermodyna Hyderabad, 1997.</li> <li>K V Narayanan, "A Textbook of Chemica Prentice Hall Of India, New Delhi 2011</li> </ol> </li> <li>Suggested References Books <ol> <li>R.C. Srivastava, "Thermodynamics an core comparison of the state of the</li></ol></li></ul>	Equilibria; Solid-Liquid Equ [6L-1	
<ol> <li>J.M. Smith, H.C. Van Ness and M.M. Abbott, "In Thermodynamics", 7th edition, McGraw-Hill Int</li> <li>Y.V.C.Rao, "Chemical Engineering Thermodyna Hyderabad,1997.</li> <li>K V Narayanan, "A Textbook of Chemica Prentice Hall Of India, New Delhi 2011</li> <li>Suggested References Books         <ol> <li>R.C. Srivastava, "Thermodynamics an core compared to the second second</li></ol></li></ol>	s, equilibrium conversion of	
Hyderabad, 1997. 3. K V Narayanan, "A Textbook of Chemics Prentice Hall Of India, New Delhi 2011 Suggested References Books 1. R.C. Srivastava, "Thermodynamics an core co	rnational Edition,2005.	eering
1. R.C. Srivastava, "Thermodynamics an core co		amics",
1. R.C. Srivastava, "Thermodynamics an core co		
deel	urse", 3 <sup>rd</sup> edition, PHI publ	ication,
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BOS held on 13th May 2019	1 10	Page 8
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गुरू घासीदास विश्वविद्यालय (केन्रीय विश्वविद्यालय अधिनेयम 2009 ज्ञ. 25 के अंतर्गत खावित केन्न्रीय विश्वविद्यालय) कोनी, बिलासपुर - 495009 (छ.ग.)



Guru Ghasidas Vishwavidyalaya (A Central University Established by the Central Universities Act 2009 No. 25 of 2009) Koni, Bilaspur – 495009 (C.G.)

**B.** Tech Syllabus

Department of Chemical Engineering

CH04TPC04 Numerical Methods in Chemical Engineering

[L:3, T:1,P:0]

15% Change

UNIT -1 Introduction of Errors and their Analysis, types of errors, numerical problems on error analysis, curve fitting: method of least squares, fittings of straight line and parabola and by method of moments, fitting of exponential curves  $y = ae^{bx}$ , fitting of the curve  $y = ab^x$ , fitting of the curve  $y = ax^b$ .

UNIT – II Numerical Solution of Algebraic and Transcendental Equations: Graphical method bisection Method, Secant Method, Regula-falsi Method, Newton Raphson Method, Solution of a system of simultaneous linear algebraic Equations Direct method: Gauss elimination Method, Gauss Jordan method, Iterative methods .Jacobi Iterative Method, Gauss Seidel Iterative method.

UNIT – III The Calculus of Finite Differences: Finite differences, Difference formula, operators and relation between operators. Inverse Operator, Interpolation with equal intervals:

Newton's forward and backward interpolation formula. Interpolation with Unequal intervals:
Lagrange's interpolation Newton's difference formula, inverse interpolation.

UNIT -IV Numerical Differentiation and Integration: - Numerical Differentiation Newton's forward and Backward difference interpolation formula. Maxima and Minima of a Tabulated function, Numerical Integration :- Trapezoidol rule, simpson is (1/3)rd and (3/8)th rule, Boole's rule, weddle rule, Difference Equations -: Definition ,order and degree of a difference equation, Linear difference equations, Difference equations reducible to Linear form simultaneous difference equations with constant coefficients.

UNIT - V Numerical solution of ordinary differential equation : Taylor series method, Euler's method, Modified Euler method Runge's method Runge Kutta method, numerical method for solution of partial differential equations. General linear partial differential equation. Laplace equation and Poisson equation.

#### **Books Recommended :**

- 1. JAIN & IYNGAR Numerical Methods for Scientific and Engineering Computations.
- 2. RAO G.S. Numerical Anlysis.
- 3. Grewal B S Numerical Methods In Engineering and Science.
- 4. Das K K Advance Engineering Methods.
- 5. Rajaraman V Computer Oriented Numerical Methods

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B. Tech Syllabus

#### Department of Chemical Engineering

#### CH04TPC06 Particle and Fluid Particle-Processing [L:3, T:0,P:0]

Pre-requisites :Fluid Mechanics

20% Change

Objectives

Objective of this course is to introduce students to the numerous industrial operations dealing with the particulate solids, their handling in various unit operations, and those in which particle fluid interactions are important. The course addresses fundamentals of fluid-particle mechanics, such as the notion of drag, and builds on those fundamentals to develop design concepts for various industrial processes like packed bed operation, fluidized operations, sedimentation, filtration, separation of solids and fluids, etc. Industrial applications are discussed. The course is concluded with an introduction to colloidal systems, soft materials and nanoparticles. Applications of these novel systems are discussed. **Contents :** 

- Introduction: Relevance of fluid and particle mechanics, and mechanical operations, in chemical engineering processes. [1L+0T]
- Solid particle characterization: Particle size, shape and their distribution, Screen analysis, standard screens; Relationship among shape factors and particle dimensions; Specific surface area; Measurement of surface area. [3L+1T]
- Mixing and storage of Solids: Types of important mixers like kneaders, dispersers, masticaters, roll mills, muller mixer, pug mixer, blender, screw mixer etc., mixing index; Types of storage equipments, Bin, Silo, Hoper, etc. [3L+1T]
- Transport of fluid-solid systems: mechanical conveying, pneumatic and hydraulic conveying. [2L+1T]
- Size reduction: Major equipment's- Crushers, grinders, ultrafine grinders, laws of communition, Close circuit and open circuit grinding. [3L+1T]
- Mechanical separations: Industrial screen; their capacity and effectiveness. [2L+1T]
- Sedimentation: Elutraition, Classification and sedimentation, Free Settling, hindered settling, flow of solids through fluid, Stoke's law, Richardson-Zaki equation, design of settling tanks. [3L+1T]

Centrifugal separation, design of cyclones and hydrocyclones. [2L+1T]

- Separation of solids from fluids: Introduction, filter bags, venture scrubber, electrostatic precipitator. [2L+1T]
- Filtration: cake filtration, Concepts, plate and frame filter, leaf filter, rotary drum filter, etc. [3L+1T]

 Fluidization: Fluidized bed, minimum fluidization velocity, pressure drop etc. Types of fluidization: Particulate fluidization, Bubbling fluidization, Applications of fluidization. [3L+1T]

 Packed bed: Void fraction, superficial velocity, channelling, Ergun equation and its derivation, Kozeny Carman equation, Darcy's law and permeability, Blaine's apparatus. [3L+1T]

13. Introduction to nanoparticles: Properties, characterization, synthesis methods, applications. [3L+1T] spin Man Total [33L+12T] BOS held on 13th May 2019 Page 12

गुरू घासीदास विश्वविद्यालय (केन्रीय विश्वविद्यालय) कोनी, बिलासपुर - 495009 (छ.ग.)



Guru Ghasidas Vishwavidyalaya (A Central University Established by the Central Universities Act 2009 No. 25 of 2009) Koni, Bilaspur – 495009 (C.G.)

	5 TO 18	ment of Chemical Engineering
CH04TPC07	Process Instrumentation []	50% Change L:3, T:0,P:0]
	[1	and a role tol
devices for temper introduced to impo	ourse is to introduce the basics of instru- n practical experience. Principles of op ature, level, pressure, flow, pH, humidity art knowledge of transmitters, transduce components related to PLC, DCS, SCADA	eration of different measuring , density, and viscosity will be
Contents :		
	ol system components, signals and standar	ds [3L+1T]
2. Pressure measu	ring instruments/sensors	[3L+1T]
<ol><li>Level measurer</li></ol>	nent	[3L+1T]
4. Flow measuring	g instruments	[3L+1T]
5. Temperature m	easuring devices	[3L+1T]
6. Humidity, dens	ity, viscosity and pH measuring devices	[3L+1T]
7. Pressure contro	llers: regulators, safety valves	[3L+1T]
8. Flow control ac	tuators; different types of valves	[3L+1T]
9. Electrical and p	meumatic signal conditioning and transmis	sion [5L+2T]
10. Computer proce	ess control, PLC, DCS, SCADA	[2L+1T]
11. Instrumentation	of process equipment	[2L+1T]
Total [33L+12T]		
Graw Hill (200: 2. S.K. Singh, Ind Suggested Referent 1. Seborg, D.E., edition, John W 2. Stephanopoulos	nn, Fundamentals of Industrial Instruments 5). ustrial Instrumentation and Control, 3 <sup>rd</sup> edi ces Books Edgar, T.F., Mellichamp, D.A.,"Process	tion, McGraw-Hill (2008). Dynamics and Control", 2 <sup>nd</sup>
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**Program Revision** 

Criteria – I (1.1.2)